



THE UNITED STATES OF AMERICA

TO ALL TO WHOM THESE PRESENTS SHALL COME:

UNITED STATES DEPARTMENT OF COMMERCE

United States Patent and Trademark Office

January 05, 2005

THIS IS TO CERTIFY THAT ANNEXED HERETO IS A TRUE COPY FROM THE RECORDS OF THE UNITED STATES PATENT AND TRADEMARK OFFICE OF THOSE PAPERS OF THE BELOW IDENTIFIED PATENT APPLICATION THAT MET THE REQUIREMENTS TO BE GRANTED A FILING DATE.

APPLICATION NUMBER: 60/527,795

FILING DATE: *December 08, 2003*

RELATED PCT APPLICATION NUMBER: *PCT/US04/40366*



Certified By

Jon W Dudas

Under Secretary
of Commerce for Intellectual Property
and Acting Director of the
United States Patent and Trademark Office

PROVISIONAL APPLICATION COVER SHEET

This is a request for filing a PROVISIONAL APPLICATION under 37 CFR 1.53 (b)(2)

Docket Number	81,641	Type a plus sign (+) inside this box →	+
---------------	--------	--	---

INVENTOR(s)/APPLICANT(s)			
LAST NAME	FIRST NAME	MIDDLE INITIAL	RESIDENCE (CITY AND EITHER STATE OR FOREIGN COUNTRY)
Nayak	Kasinath		2 Louisiana Cove, Odessa, TX 79762
Cummings	Gerald		4230 Lazy Creek Drive, Tyler, TX 75707
Merrill	Roger		4309 Kirkwood Drive, Odessa, TX 79762

TITLE OF THE INVENTION (280 characters max)

RADIATION RESISTANT POLYPROPYLENE USEFUL IN MEDICAL APPLICATIONS
--

CORRESPONDENCE ADDRESS

Legal Department Huntsman LLC P.O. Box 15730
--

STATE	Austin, Texas	ZIP CODE	78761	COUNTRY	USA
-------	---------------	----------	-------	---------	-----

ENCLOSED APPLICATION PARTS (check all that apply)

<input checked="" type="checkbox"/> Specification	Number of Pages <u>24</u>	<input type="checkbox"/> Small Entity Statement
<input type="checkbox"/> Drawing(s)	Number of Sheets <u> </u>	<input checked="" type="checkbox"/> Other (specify) <u>Postcard</u>

METHOD OF PAYMENT (check one)

<input type="checkbox"/> A check or money order is enclosed to cover the Provisional filing fees	PROVISIONAL FILING FEE AMOUNT	\$ 160.00
<input checked="" type="checkbox"/> The Commissioner is hereby authorized to charge the filing fee to Deposit Account Number: <u>08-3442</u>		

The invention was made by an agency of the United States Government or under a contract with an agency of the United States Government.

☒ No.
☐ Yes, the name of the U.S. Government agency and the Government contract number are:

Respectfully submitted,

SIGNATURE

Christopher J. Whewell

Date

12/8/03

NAME: CHRISTOPHER J. WHEWELL

Reg. No.

37,469

☐

Additional inventors are being named on separately numbered sheets attached hereto.

Certificate of Mailing By Express Mail

I hereby certify that this correspondence is being deposited with the U.S. Postal Service Express Mail Service No. EV165093790US addressed to Commissioner for Patents, Alexandria, VA 22313 on December 8, 2003.

Martha Victory
Martha Victory

PROVISIONAL APPLICATION FILING ONLY

Radiation Resistant Polypropylene Useful in Medical Applications

5

Field of the Invention

This invention relates to the development of an improved clear, color-stable, and radiation-sterilizable polypropylene for various medical applications, including syringes and other wares normally fabricated from what is recognized in the art as radiation-sterilizable polyolefins.

10

Background Information

15

US patent #4,666,959 teaches the use of a polymeric hindered amine light stabilizer, an alkyl phosphite and a specific phenolic antioxidant as necessary additives to protect polypropylenes from the exposure to high energy "gamma" radiation. US patent #4,888,369 teaches very similar art as the previous one, except it teaches the use of an additive termed as mobilizer, such as hydrocarbon oil. Although the teachings of these patents are helpful to prevent degradation of polypropylenes from exposure to high energy radiation, the main drawbacks are the yellowing of parts made from the teachings therein due to the presence of hindered phenolic antioxidants. US Patent #6,231,936 claims the radiation tolerant polypropylene composition comprising of polypropylene and polyethylene (1-50%) produced by single-site catalyst along with additives such as hindered amine stabilizers, secondary antioxidants (i.e., phosphites and thioesters), sorbitol type clarifiers. However secondary antioxidants such as phosphites are susceptible to hydrolysis upon exposure to moisture prior to or during extrusion. Also, in the real world case scenario, most of

20

25

resin which show up in the molded parts thereafter. This patent also claims the use of thiodipropionate secondary antioxidant selected from the group consisting of distearyl thiopropionate and dilaurylthiopropionate. These sulfur containing additives are known to impart odor. This patent also claims the addition of a sorbitol type clarifying agent (i.e., bis-4-methylbenzylidene sorbitol and bis-3,4-dimethylbenzylidene sorbitol) up to 0.5% to enhance clarity of molded parts. However, sorbitol type clarifying agents impart cherry flavored odor, and in some severe autoclave conditions (per 9 CFR 121 condition A), they form flocculates.

The presentation "New Improvements in Radiation Resistant Polypropylenes" presented at the Fifth International Conference Additives in 1996 showed the formulations consisting of special additives combinations of hindered amine light stabilizers ("HALS") and phosphites improved the color-stability of polypropylenes after exposure to gamma radiation.

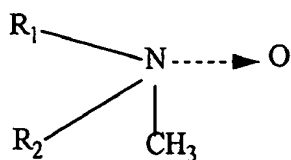
The polypropylenes commonly contain a hindered phenolic antioxidant along with a secondary antioxidants such phosphites and thioesters as the processing stabilizers. However, upon exposure to gamma radiation, it exhibits undesirable color due to generation of color-bodies from the oxidized hindered phenolic type primary antioxidants. A non-hindered phenolic additive system for such applications is desired. Although addition of a phosphite prevents the discoloration, the phosphite is also the main source of causing black specks in the products during the end-use applications. Although thioesters are good processing as well as thermal stabilizers, they do impart odor, which is not acceptable for most medical uses. Hence, this invention relates to an additive composition that is free of both hindered phenolics, and secondary antioxidants such as phosphites and thioesters. It comprises a combination of a

hindered amine light stabilizer and an amine oxide, or a combination of hindered amine light stabilizer and hydroxyl amine compounds that provided excellent color stability after exposure to gamma radiation up to 5 mrads. Also addition of clarifier such as NA-21 (Amfine Chemical Corporation) imparted excellent clarity.

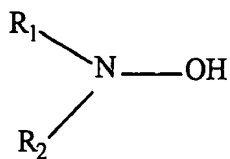
The present invention remedies the aforementioned deficiencies by achieving better color stability along with enhanced clarity and impact resistance. A polypropylene random copolymer (nominal MFRs and 9 and ~ 25 dg/min @ 230C/2.16kg per ASTM D-1238) consisting of I) a combination of hindered amine light stabilizer ("HALS") and an amine oxide along with an acid neutralizer (i.e., metallic stearate) or II) a combination of HALS and hydroxyl amine along with an acid neutralizer was exposed to gamma radiation up to 5 mrads. The results indicated that HALS/amine oxide or HALS/ hydroxylamine maintained excellent color stability, even showing very little increase in yellowness index after exposure to gamma radiation. Addition of a new clarifier NA-21 (Amfine, Allendale, NJ) imparted better clarity having less effect on tensile and impact properties unlike sorbitol based clarifier such as MILLAD® 3988 (Milliken Chemical Co, Spartanburg, South Carolina). Addition of a metallocene catalyzed polyethylene polymer and Ziegler-Natta catalyzed polyethylene containing octene as a comonomer improved the impact strength of the polymer after being exposed to gamma radiation.

Summary of the Invention

The present invention provides a blend useful as an additive in polyolefin polymers for minimizing the effects of radiation on the physical properties of said polymers, which comprises a hindered amine light stabilizer and at least one material selected from the group consisting of: i) amine oxides exemplified by the formula:



in which R_1 and R_2 are each independently selected from C_{10} to C_{24} alkyl, aryl, or alkylaryl groups, whether straight-chain, branched, cyclic, saturated, or unsaturated; and ii) hydroxylamines exemplified by the formula:



in which R_1 and R_2 are each independently selected from C_{10} to C_{24} alkyl, aryl, or alkylaryl groups, whether straight-chain, branched, cyclic, saturated, or unsaturated.

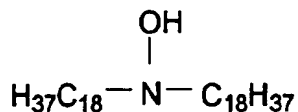
Detailed Description

Semi-crystalline polymers such as polypropylene are used in medical devices, and food packaging where these articles are frequently subjected to ionizing radiation for sterilization. It is known that exposure of polymers such as polypropylene to high-energy radiation i.e., electron beam or gamma radiation triggers radiation induced chemical reactions with predominant chain scission mechanism, resulting in loss of physical properties. Such property losses include embrittlement and discoloration, and are not acceptable to end-use applications.

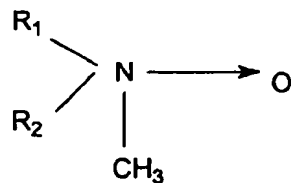
The present invention provides novel formulations of polypropylene compositions that are radiation resistant, color-stable and clear. In accordance with the present invention, polypropylene and propylene-ethylene copolymer compositions comprise of the following:

- i. 0.01 - 0.2 wt% hindered amine light stabilizer ("HALS")
- ii. 0.01 - 0.1 wt% amine oxide
- iii. 0.01 - 0.2 wt% hydroxyl amine
- iv. 0.01 - 0.3 wt% clarifier or nucleator
- v. 0.01 - 0.2 wt% acid neutralizer

The general chemical formulae for amine oxides and hydroxyl amines are illustrated as:

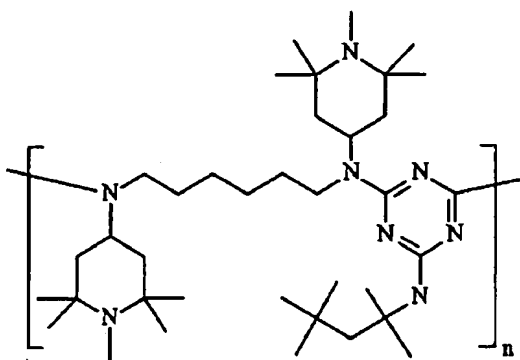


Hydroxyl Amine (IRGASTAB® FS-042)
(Ciba Specialty Chemicals)



$\text{R}_1, \text{R}_2 = \text{C}_{14}\text{-C}_{24}$ alkyl chain

Amine Oxide (GENOX® EP)
(Crompton Corporation)



Hindered Amine Light Stabilizer, CHIMASSORB® 944, (Ciba Specialty Chemicals)

A combination (1:2) of amine oxide (GENOX® EP) and HALS or 1:1 blend of hydroxyl amine (FS-042) with HALS (CHIMASSORB® 944) more commonly known as IRGASTAB® FS 410 were thoroughly tested, and compared to those obvious ones known from the prior art. The individual additives in various formulations are identified in the Additive Index below.

1. Naugard XL-1 (CAS #70331-94-1, 2,2'-oxiamidobisethyl 3(3,5-di-tert-butyl-4-hydroxyphenyl)propionate, Crompton Corporation)
2. TINUVIN® 622LD (CAS #65447-77-0, Dimethyl succinate polymer with 4-hydroxy-2,2,6,6-tetramethyl-1-piperidineethanol; Ciba Specialty Chemicals)
3. CHIMASSORB® 944LD (CAS #71878-19-8;N, N'-bis(2,2,6,6-tetramethyl-4-piperidiny)-1,6-hexanediamine,polymer with 2,4,6-trichloro-1,3,5-triazine and 2,2,4-trimethyl-1,2-pentaamine, Ciba Specialty Chemicals)
4. GENOX® EP (CAS #204933-93-7; Dialkyl methyl amine oxide; Crompton Corporation)
5. IRGASTAB ® FS410 (1:1 blend of IRGASTAB® FS042 and CHIMASSORB® 944LD from Ciba Specialty Chemicals)
6. DHT-4A (CAS # 11097-59-9; Synthetic hydrotalcite; Kyowa Chemical)
7. Calcium Stearate (CAS # 1592-23-0;Crompton Corporation)
8. PEP-36 (CAS #80693-00-1;Bis (2,6-di-tert-butyl-4-methylphenyl)pentaerythritol-di-phosphite; Amfine Chemical)
9. ULTRANOX® 641 (CAS #161717-32-4; 2,4,6-tri-tert-butylphenyl 2, butyl 2ethyl 1,3prpane diol phosphite; Crompton Corporation)
10. WESTON® 619 (CAS #3806-34-6, Distearyl Pentaerythritol Diphosphite; Crompton Corporation)

Additives for improving clarity and nucleation are as follows.

1. MILLAD® 3988 (CAS # 135861-56-1; Bis 3,4,-dimethylbenzylidene sorbitol, from Milliken Chemical)
2. HPN-68 (CAS # 351870-33-2, Proprietary inorganic salt from Milliken Chemical)
3. NA-21 (Proprietary inorganic salt from Amfine Chemical)
4. NA-11 (CAS #85209-91-2, Sodium 2,2'-methylene-bis-(4,6-di-tert-butylphenyl)phosphate; from Amfine Chemical)
5. KM-1500 (CAS #1309-43-4 and 68440-56-2; Magnesium Salt of disproportionated rosin acid ; Mitsui&Co)

Impact Modifiers for improving impact properties are as follows.

1. Engage 8200 (CAS #026221-73-8); Single site catalyzed polyethylene; DuPont Dow Chemical)
2. LL-1002.09 (CAS #25087-34-7, Zigler Natta Catalyzed Linear Low Density polyethylene with butene comonomer; ExxonMobil Chemicals)
3. L8101 (CAS #26221-73-8, Zigler Natta Catalyzed Linear Low density polyethylene with octene comonomer; Huntsman Polymers)

An unexpected result of the invention includes improved color stability, enhanced clarity, and improved impact resistance after being exposed to gamma radiation without use of conventional phosphite and thioesters.

Example 1

The formulations A1 through N1 are given in Table 1. Several formulations had hindered amine light stabilizer (HALS)/phosphites. The phosphites used are ULTRANOX® 641 (Crompton Corporation, Middlebury, CT), WESTON® 619 (Crompton Corporation, Middlebury, CT), DOVERPHOS® S-9228T (Dover Chemical, Dover, OH), PEP-36 (Amfine, Allendale, NJ). Also, clarifiers/nucleators such as MILLAD® 3988 (Milliken Chemical Spartanburg, SC), HPN-68 (Milliken Chemical, Spartanburg, SC), NA-21 (Amfine Chemical, Allendale, NJ), and KM1500 (Mitsui Chemical) were added at specified levels. The hindered amine light stabilizers are TINUVIN® 622 and CHIMASSORB® 944 (Ciba Specialty Chemicals, Tarrytown, NY). The amine oxide is GENOX® EP (Crompton Corporation, Middlebury, CT.). The impact modifier chosen was ENGAGE® 8200 (a metallocene-catalyzed ethylene copolymer from Dupont-Dow). The specific formulations were pre-blended with unstabilized polypropylene random copolymer powder (MFR ~ 12 g/10min), and then melt extruded by a Haake TW100 twin screw extruder at a processing temperature of 230°C. Test specimens were molded on a 120-ton Van Dorn injection-molding machine under ASTM Conditions. Specimens were irradiated at 2.0 and 4.0 mrad by Isomedix, Whippany, NJ using a ⁶⁰Co source. Control specimens (i.e., non-radiated) were included in each physical test for comparison. Yellowness Index was measured with a Colorgard/05 from BYK Gardner as per ASTM D-1925. Tensile properties were measured on Typ1-I injection molded tensile bars with an Instron 1125 universal testing machine with an initial grip separation of 2.5" and an extension rate of 5"/min. Melt flow rate (MFR) was measured with a Kayness Galaxy I melt indexer as per

ASTM-1238B. Multiaxial impact energy was measured by Dynatup 8250, using velocity of 4.3m/sec and crosshead weight of 28 lbs (12.7kg).

Effect of Gamma Radiation on MFR: The % change in MFR after exposing the samples at 2 and 4 mrads are given in table 1. Sample K1 (containing CHIMASSORB® 944 @ 0.15% and ULTRANOX® 641 @ 0.1%) had the least increase of 316.7% in MFR after 2 mrads of exposure, whereas sample F1 (control - containing 0.2% TINUVIN® 622 + 0.1% WESTON® 619) had the highest increase of 671.7%. At 4.0 mrads of exposure, sample M1 (0.15% CHIMASSORB® 944+0.1% PEP-36 + 2.5% ENGAGE® 8200) had the least increase of MFR, whereas Sample F1 (Control sample) had the highest increase in MFR of 1704.3%. Sample J1 (containing CHIMASSORB® 944 @ 0.1%, and GENOX® EP @ 0.05%) had the corresponding increase of 325% and 741.7% at 2.0 and 4.0 mrads of exposure, showing that it was relatively more stable compared to the control sample F1.

Table 1							
Sample ID	A1	B1	C1	D1	E1	F1	G1
PP Powder (MFR ~10 dg/min), %	99.575	99.525	99.495	99.425	99.495	99.23	99.475
Chimassorb 944, wt%	0.15	0.15	0.15	0.15	0.15		0.15
Tinuvin 622, wt%						0.2	
Ultranox 641, wt%	0.1	0.1	0.1	0.1	0.1		0.1
Doverphos S-9228T, wt%							
Genox EP, wt%							
Calcium Stearate, wt%	0.05	0.05	0.05	0.05	0.05	0.1	0.05
DHT-4A, wt%	0.025	0.025	0.025	0.025	0.025	0.05	0.025
Naugard XL-1, wt%						0.07	
Weston 619, wt%						0.1	
NA-21, wt%	0.1	0.15	0.18	0.1			
Millad 3988, wt%				0.15	0.18	0.25	0.15
NA-11, wt%							0.05
Physical Properties at 0.0 Mrads							
MFR (g/10min)	12	11	11	11	11	9.2	12
% Strain at Yield	13.5	13.6	13.9	14	13.9	12.4	14.1
Stress at Yield, psi	4200	4220	4220	4270	4350	4570	4340
%Strain at Break	560	580	570	610	660	650	650
Stress at Break, psi	2760	2770	2780	2770	2850	2910	2860
Multiaxial Impact @ 23°C, in-lbs	250(112)	278(114)	254(73)	52(29)	27(13)	21(4)	157(84)
Yellowness Index	4.5	4	3.9	4.6	4.8	4.2	4.8
%Haze-25mil	8.5	7.8	7.3	8.3	8.6	10.7	9.1
%Haze-50mil	23.2	21.6	20.7	22.5	21	22.6	21.1
Crystallization Temp (°C)	118.3	119.6	119.4	119.7	119.9	120.9	121
at 2.0 Mrads							
MFR (g/10min)	51	55	56	63	59	71	55
% Strain at Yield	13.6	13.7	13.7	14.1	13.8	11.9	14
Stress at Yield, psi	4170	4190	4190	4230	4320	4570	4350
%Strain at Break	630	620	620	440	510	340	570
Stress at Break, psi	2720	2740	2730	2670	2760	2620	2770
Multiaxial Impact @ 23°C, in-lbs	111	167	131	24	27	22	93
Yellowness Index	4.6	4.4	4.5	5.2	5.8	6.2	6.5
4.0Mrads							
MFR (g/10min)	147	118	129	113	127	166	137
% Strain at Yield	14.4	14	14.2	13.6	14.2	13.1	13.9
Stress at Yield, psi	4150	4200	4190	4330	4470	4590	4400
%Strain at Break	690	660	640	330	340	170	270
Stress at Break, psi	2720	2690	2690	2640	2670	3410	2690
Multiaxial Impact @ 23°C, in-lbs	55	85	43	18	15	15	35
Yellowness Index	5.4	4.8	5.4	5.7	7	6.5	7
%Change in MFR (@ 2 Mrads)	325	400	409.1	472.7	436.4	671.7	358.3
%Change in MFR (@4 Mrads)	1125	972.7	1072.7	927.3	1054.5	1054.5	883.3
Diff. In YI (2.0-0 Mrads)	0.1	0.4	0.6	0.6	1	2	1.7
Diff. In YI (4.0-0 Mrads)	0.9	0.8	1.5	1.1	2.2	2.3	2.2

Effect of Nucleator/Clarifiers: Since, gamma radiation on clarity of PP has negligible effect, the step-plaques (25/50 mils) were not subjected to gamma radiation. The % haze (for 25 mil plaque) of samples A1, B1, and C1 (containing NA-21 @ 0.1, 0.15, and 0.18% respectively) were 8.5, 7.8, and 7.3 respectively, showing very excellent enhancement in clarity. The corresponding % haze for 50 mil plaques were 23.2, 21.6, and 20.7 respectively, showing excellent clarity. % Haze of samples E1 and L1 (Duplicate run - both containing Millad 3988 @ 0.18%) were 8.6 and 6.3 respectively; showing some variability. However, this might be attributed to the dispersion of MILLAD® 3988 (i.e., poor dispersion may cause higher % haze; better dispersion may cause lower % haze). Hence in a reactor grade PP, both NA-21 and MILLAD® 3988 imparted similar enhancement in clarity (i.e., reduction in %haze). Samples M1 and N1 (containing 2.5% and 5% ENGAGE® 8200) had the %haze (25 mil) of 8.2 and 9.3 respectively. This indicated that the modifier (i.e., ENGAGE® 8200) had very little effect on clarity. Combination of MILLAD® 3988/HPN-68 (@ 0.15%/0.05%) in samples H1, J1 and I1 imparted % haze values of 9.2, 7.4, and 7.5 units, measured on 25 mil thick plaques. Also, these three samples had crystallization temperatures almost 2° higher than that of sample C1 (containing NA-21 @ 0.18%). Higher crystallization temperature normally results in lower cycle time, and higher productivity.

Effect of Gamma Radiation on Yellowness Index: The initial colors of samples A1 through N1 were in the range of 2.1 – 5.4 units. Sample H1 had the highest initial color of 5.4. The control sample F1 had the initial color of 4.2 units, 6.2 units @ 2 mrads, and 6.5 units @ 4 mrads, thus an increasing trend in yellowness index with increase of dosage of gamma radiation.

Sample I1 (0.15% C-944+ 0.1% DOVERPHOS® S-9228T) had initial color of 4.3 @ 0.0 mrads, 3.9 units @ 2.0 mrads, and 3.9 units @ 4.0 mrads, thus showing slight decrease in yellowness index with increase in dosage of gamma radiation. Sample J1 (0.1% C-944+0.05% GENOX® EP had initial color of 4.6units, 2.3 units @ 2 mrads, and 2.6 units @ 4 mrads, thus showing decrease in yellowness index with increase in dosage of gamma radiation. All other formulations showed an increase in yellowness index with increase of the dosage of gamma radiation.

Table 1 (Continued)							
Sample ID	H1	I1	J1	K1	L1	M1	N1
PP Powder (MFR ~10 dg/min), wt%	99.475	99.475	99.575	99.475	99.495	96.875	94.375
Chimassorb 944, wt%	0.15	0.15	0.1	0.15	0.15	0.15	0.15
Tinuvin 622, wt%							
Ultranox 641, wt%	0.1			0.1	0.1	0.1	0.1
Doverphos S-9228T, wt%		0.1					
Genox EP, wt%			0.05				
Calcium Stearate, wt%	0.05	0.05	0.05	0.05	0.05	0.05	0.05
DHT-4A, wt%	0.025	0.025	0.025	0.025	0.025	0.025	0.025
Naugard XL-1, wt%							
Weston 619, wt%							
NA-21, wt%							
Millad 3988, wt%	0.15	0.15	0.15	0.15	0.18	0.15	0.15
NA-11, wt%							
HPN-68, wt%	0.05	0.05	0.05			0.05	0.05
KM-1500, wt%				0.05			
PEP-36, wt%						0.1	0.1
Exact8200, wt%						2.5	5
Physical Properties							
at 0.0 Mrads							
MFR (g/10min)	12	12	12	12	12	11	12
% Strain at Yield	14.3	14.2	14.2	14.3	13.9	14.9	15.4
Stress at Yield, psi	4290	4250	4260	4320	4360	4020	3880
%Strain at Break	630	610	620	620	700	590	630
Stress at Break, psi	2790	2800	2810	2820	2850	2750	2740
Multiaxial Impact @23C, in-lbs	257(106)	301(61)	285(73)	278(72)	318(49)	334(5)	325(3)
Yellowness Index	5.4	4.3	4.6	4.6	5.9	3.7	2.1
%Haze-25mil	9.2	7.4	7.5	8.1	6.3	8.2	9.3
%Haze-50mil	21.6	19.6	18.9	20.5	14.3	20.6	23.7
Crystallization Temp (C)	121.5	122.5	122.5	120.1	120.9	121.9	122
at 2.0 Mrads							
MFR (g/10min)	58	62	51	50	54	47	57
% Strain at Yield	14.2	14.2	14.2	13.8	14.1	14.5	15.3
Stress at Yield, psi	4280	4260	4280	4310	4400	4080	3880
%Strain at Break	480	320	640	530	670	580	580
Stress at Break(Psi)	2700	2680	2800	2770	2870	2720	2690
Multiaxial Impact @ 23C, in-lbs	131	106	201	160	155	291	307
Yellowness Index	6.9	3.9	2.3	4.7	5.1	6.7	5.7
at 4.0 Mrads							
MFR (g/10min)	118	107	101	129	116	91	117
% Strain at Yield	13.6	14	14.3	13.7	14	14.5	15.1
Stress at Yield(psi)	4300	4290	4270	4320	4380	4060	3880
%Strain at Break	280	270	290	300	340	370	570
Stress at Break, psi	2670	2640	2640	2650	2690	2610	2610
Multiaxial Impact @ 23C, in-lbs	58	45	37	50	31	243	294
Yellowness Index	7.7	3.9	2.6	6	6.2	5.1	4.5
% Change in MFR (@2 Mrads)	383.3	416.7	325	316.7	350	327.3	375
% Change in MFR (@4 Mrads)	883.3	791.7	741.7	975	866.7	727.3	875
Diff. In YI (2.0 -0.0 Mrads)	1.5	-0.4	-2.3	0.1	-0.8	3	3.6
Diff. In YI (4.0 -0.0 Mrads)	2.3	-0.4	-2	1.4	0.3	1.4	2.4

Effect of Gamma Radiation on Tensile Properties: % Strain @ yield was unaffected by increase in dosage of gamma radiation. % Strain @ break was affected to some degree with increase in dosage of gamma radiation. At 2.0 mrads of exposure, the losses in % strain at break of various formulations were in the range of 1.7% - 47.5%; Sample F1 had the highest loss of % strain at break. M1 (containing 2.5% ENGAGE® 8200) had 1.7% loss of strain at break at 2.0 mrads and 37.3% loss at 4mrads of exposure respectively. Sample N1 (containing 5% ENGAGE® 8200) had the corresponding loss of 7.9% and 9.5% at 2 and 4 mrads respectively. Hence, it is evident that addition of ENGAGE® 8200 (@ 2.5 – 5%) helped to maintain the % strain at break.

Effect of Gamma Radiation on Multiaxial Impact: Multi-axial impact of samples A1 through N1 were measured by 8250 Dynatup with an impact of 26 lbs under acceleration due to gravity. The initial impact values varied from 21 in-lbs to 334 in-lbs. It was evident that the samples containing MILLAD® 3988 had the least impact values compared to the samples containing NA-21. The loss of impact at 2 mrads were in the range of 0 – 55.6%. The loss of impact at 4 mrads were in the range of 9.5 – 89.4%, showing a higher degree of loss at this dosage of gamma radiation. Sample M1 (containing 2.5% ENGAGE® 8200) had a loss of ~27% in impact; whereas sample N1 (containing 5% ENGAGE® 8200) had only 9.5% loss of impact. Hence, it is evident that addition of ENGAGE® 8200 helped to maintain impact properties at higher dosage of gamma radiation.

Example 2

In the second example, some of the formulations of example-1 were repeated. Here the base resin chosen was a random copolymer polypropylene, having melt flow rate of 25 gm/10min. The formulations (A2-C2) are given in Table-2. Sample B2 contained a modifier "ENGAGE® 8200" @ 2.5%, and sample C2 contained a linear low density polyethylene available from Huntsman Polymers Corporation of Odessa, Texas under the tradename of "LLDPE L8101", which is an octene copolymer. All these formulations were pre-blended with an un-stabilized random copolymer having MFR ~ 25 g/10min, and then were compounded by a 2.5" Davis Standard single screw extruder at a processing temperature (210°C). The test specimens (prepared - as described in example-1) were irradiated at 2.5 and 5.0 mrads of gamma radiation by Isomedix, Whippany, NJ. All these samples were tested as described in example 1.

The yellowness indices of these formulations had very minimal increase at 5 mrads of exposure vs. those of non-radiated samples, showing excellent color stability. Samples C2 had relatively lower increase in MFR at 5.0 mrads. This could be attributed to the presence of a LLDPE (L8101 @5wt%) resulting in crosslinking. Addition of impact modifier such as ENGAGE® 8200 (even @ 2.5%) resulted in better impact energy than the control sample. Addition of L8101 @ 5wt% did not provide much improvements in impact resistance.

Table -2			
Sample ID	A2	B2	C2
PP Powder (MFR -25dg/min),wt%	99.55	97.05	94.55
Chimassorb 944,wt%	0.15	0.15	0.15
Genox EP, wt%	0.05	0.05	0.05
Calcium Stearate,wt%	0.05	0.05	0.05
DHT-4A, wt%	0.025	0.025	0.025
Engage 8200, wt%		2.5	
NA-21, %	0.175	0.175	0.175
LB101, wt%			5
Physical Properties			
at 0.0 Mrads			
MFR (g/10min)	26.8	26.9	23.8
%Strain at Yield	13.8	14.2	14
Stress at Yield,psi	4010	3800	3830
%Strain at Break	650	640	370
Stress at Break, psi	2590	2500	2490
% Haze - 25 mil	9.8	12.1	20.5
%Haze - 50 mil	27.9	33.5	44.1
Yellowness Index	-0.11	-2.4	-1.44
Multiaxial Impact at 23C, in-lbs	60	121	66
at 2.5 Mrads			
MFR (g/10min)	127	134	117
%Strain at Yield	13.9	14.7	14.2
Stress at Yield,psi	4020	3810	3830
%Strain at Break	480	640	370
Stress at Break, psi	2480	2480	2490
% Haze - 25 mil	9.9	12.1	20.5
%Haze -50 mil	28	33.2	44.3
Yellowness Index	0.72	-1.25	0.38
Multiaxial Impact at 23C, in-lbs	52	91	42
at 5.0 Mrads			
MFR (g/10min)	365	339	230
%Strain at Yield	14.4	15	15
Stress at Yield,psi	3990	3810	3810
%Strain at Break	360	460	590
Stress at Break, psi	2380	2370	2410
% Haze - 25 mil	10	12.3	20.6
%Haze -50 mil	27.8	33.6	44.3
Yellowness Index	1.29	0.49	0.75
Multiaxial Impact at 23C, in-lbs	20	54	26

Example 3

In this example, we have repeated some formulations as given in example 1 and also included FS410 as a primary stabilizer system (see Samples F3 and G3). The control formulation was sample D3. The additive formulations (shown in Table-3) with an un-stabilized polypropylene powder having initial MFR ~ 25 dg/min were pre-blended and compounded by Haake TW100 twin screw extruder. The test specimens were irradiated at 2.5 and 5.0 mrads as done previously. The test specimens were tested as described in prior example 1.

The control sample D3 had YI colors of -1.79 (@ 0.0 mrads), 5.39(@2.5 mrads), and 5.91 (@5.0 mrads). The samples B3 (containing CHIMASSORB® 944 and GENOX® EP along with ENGAGE® 8200) had the least YI value of -1.67 after being exposed to 5 mrads of gamma radiation. Also the samples (F3 and G3) containing FS410 exhibited much lower YI (1.09 and 0.54 respectively) at 5 mrads' exposure. Hence, it is apparent that additive formulations consisting of either a combination of amine oxide and a HALS (i.e., CHIMASSORB® 994) or FS410 exhibited excellent color stability after being exposed to gamma radiation up to 5 mrads.

Table - 3							
Sample ID	A3	B3	C3	D3	E3	F3	G3
PP Powder(MFR-25dg/min),wt%	99.625	94.625	99.475	94.23	99.575	99.575	99.625
Chimassorb 944,wt%	0.15	0.15	0.15		0.15		
Genox EP,wt%	0.05	0.05	0.05		0.05		
FS410,wt%						0.2	0.2
CaSt,wt%	0.05	0.05	0.05	0.1	0.05	0.05	0.05
DHT-4A,wt%	0.025	0.025	0.025	0.05	0.025	0.025	0.025
HPN-68,wt%	0.1	0.1			0.05	0.05	0.1
Millad 3988,wt%			0.25	0.25			
NA-21,wt%					0.1	0.1	
Engage 8200,wt%		5%					
Tin622LD,wt%				0.2			
NaugardXL-1,wt%				0.07			
Weston619,wt%				0.1			
LL-1002.09,wt%				5%			
Physical Properties							
at 0.0 Mrads							
MFR (g/10min)	29	28	30	27	30	30	31
Crystallization Temp (°C)	125.8	125.1	121.9	122.3	125.8	125.1	121.9
%Strain at Yield	13	14.3	13.7	12	14.1	13.9	12.9
Stress at Yield, psi	4300	3890	4540	4440	4270	4240	4380
%Strain at Break	630	440	730	370	720	770	690
Stress at Break,psi	2670	2560	2790	2650	2680	2770	2850
Multiaxial Impact @23°C,in-lbs	42	225	29	80	61	40	29
Yellowness Index	-0.98	-3.67	0.36	-1.79	-0.34	0	-0.73
%Haze - 25 mil	10.7	15.2	5.8	16.8	8.9	9.2	11.4
%Haze- 50 mil	24.7	35.8	13.6	32.8	22.4	23	24.3
at 2.5 Mrads							
MFR(g/10min)	127	123	129	152	129	112	107
Crystallization Temp (°C)	125.7	125.4	121.7	122.7	123.8	122.8	125.6
%Strain at Yield	13.4	14.3	13.6	11.9	13.8	13.2	12.6
Stress at Yield, psi	4280	3830	4540	4470	4270	4340	4350
%Strain at Break	200	300	460	320	610	600	610
Stress at Break, psi	2500	2470	2650	2550	2570	2540	2540
Multiaxial Impact @23°C,in-lbs	26	77	22	32	33	30	30
Yellowness Index	-0.13	-2.15	1.24	5.39	0.57	0.46	-0.27
at 5 Mrads							
MFR(g/10min)	223	221	234	321	228	225	242
Crystallization Temp (°C)	125.4	124.9	121.8	121.2	123.1	121.6	124.9
%Strain at Yield	12.6	14.1	13.5	11.7	13.9	13.7	13.5
Stress at Yield, psi	4400	3820	4540	4460	4270	4250	4240
%Strain at Break	520	320	240	340	210	200	190
Stress at Break, psi	2540	2440	2560	2540	2420	2450	2470
Multiaxial Impact @23°C, in-lbs	16	50	16	24	24	21	28
Yellowness Index	0.38	-1.67	1.77	5.91	1.09	1.09	0.54

Example 4

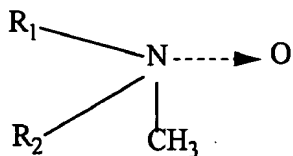
In this example, we have repeated some formulations as given in example 1 using a polypropylene copolymer having MFR ~ 10 dg/min and also included FS410 as a primary stabilizer system (see Sample F4 in Table 4). The additive formulations with un-stabilized polypropylene powder having initial MFR ~ 10 dg/min were pre-blended and compounded by Haake TW100 twin screw extruder. The test specimens were irradiated at 2.5 and 5.0 mrads of gamma radiation as done previously. The test specimens were tested as described in prior example 1.

The properties of these formulations (sample A4 - F4) are given in Table-4. The sample B4 (containing CHIMASSORB® 944/GENOX® EP plus NA-21) had YI of 0.47 units (@ 0.0 mrads), 1.3 units (@2.5 mrads) and 1.76 units at (@5.0 mrads). Sample F4 (containing FS410 and NA-21) had the corresponding YI of 1.34, 2.44 and 2.52 units. Thus it is again evident that CHIMASSORB® 944 with either GENOX® EP (amine oxide) or FS042 (a hydroxyl amine) provided excellent color stability at 5 mrads of gamma radiation. Also, addition of MILLAD® 3988 (a sorbitol based nucleator/clarifier in sample C4) showed slight increase in YI compared to other samples (i.e., sample A4 with HPN-68, Sample B4 with NA-21). Note that the specific concentrations of nucleators/clarifiers were chosen, because they are known to improve clarity/nucleation at these levels. It was also evident that addition of 10% ENGAGE ® 8200 (i.e., sample D4) resulted in higher multiaxial energy, compared to other formulations.

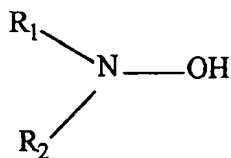
Table - 4						
Sample ID	A4	B4	C4	D4	E4	F4
PP Powder(MFR~9dg/min),wt%	99.625	99.575	99.55	89.525	99.5	99.575
Chimassorb 944,wt%	0.15	0.15	0.15	0.15	0.15	
Genox EP,wt%	0.05	0.05	0.05	0.05	0.05	
FS410,wt%						0.2
Calcium Stearate,wt%	0.05	0.05	0.05	0.05	0.05	0.05
HPN-68,%	0.125			0.075	0.1	
NA-21,wt%		0.175				0.175
Millad 3988,wt%			0.2	0.15	0.15	
Engage 8200,wt%				10%		
Physical Properties						
at 0 Mrads						
MFR(g/10min)	11	11	11	12	11	11
%Strain at Yield	13.3	14.3	14.4	16.6	13.9	14.3
Stress at Yield,psi	3990	4060	4110	3360	4240	4240
%Strain at Break	670	660	620	830	660	650
Stress at Break, psi	2650	2680	2700	2500	2710	2710
Multiaxial Impact @23°C, in-lbs	117	346	190	319	140	96
Yellowness Index	0.47	0.5	1.52	-1.65	1.13	1.34
%Haze - 25 mil	24.7	7.9	5.4	19.5	11.5	6.3
%Haze - 50 mil	47.4	19.4	10.6	46.2	23.5	13.1
at 2.5 Mrads						
MFR(g/10min)	63	66	71	56	55	54
%Strain at Yield	12.7	13.9	14.8	16.6	14.3	14.3
Stress at Yield, psi	4050	4150	4040	3370	4040	4170
%Strain at Break	610	580	530	810	660	670
Stress at Break, psi	2550	2600	2630	2430	2630	2620
Multiaxial Impact @23°C, in-lbs	81	215	136	311	80	82
Yellowness Index	1.3	1.52	2.55	0.69	2.35	2.44
at 5 Mrads						
MFR(g/10min)	72	115	95	99	103	103
%Strain at Yield	13.56	14.3	14.3	16.1	13.8	14.8
Stress at Yield, psi	3930	4040	4180	3520	4180	4080
%Strain at Break	460	650	630	880	660	540
Stress at Break, psi	2540	2440	2560	2430	2580	2560
Multiaxial Impact @23°C, in-lbs	29	128	36	301	30	42
Yellowness Index	1.76	1.86	2.87	0.59	2.21	2.52

What is claimed is:

1) A blend useful as an additive in polyolefin polymers for minimizing the effects of radiation on the physical properties of said polymers, which comprises a hindered amine light stabilizer and at least one material selected from the group consisting of: i) amine oxides exemplified by the formula:



in which R_1 and R_2 are each independently selected from C_{10} to C_{24} alkyl, aryl, or alkylaryl groups, whether straight-chain, branched, cyclic, saturated, or unsaturated; and ii) hydroxylamines exemplified by the formula:



in which R_1 and R_2 are each independently selected from C_{10} to C_{24} alkyl, aryl, or alkylaryl groups, whether straight-chain, branched, cyclic, saturated, or unsaturated.

- 2) A polymerized olefin polymer comprising the blend of claim 1 present in any amount between about 500 ppm and 5000 ppm by weight based on the total weight of said polymer.
- 3) An olefin polymer according to claim 2 wherein said polymer is selected from the group consisting of: propylene homopolymers, propylene co-polymers, ethylene homopolymers, and ethylene co-polymers, wherein when said olefin polymer comprises a co-polymer of either propylene or ethylene, said co-polymer is a co-polymer which was formed in the presence of at least one monomer comprising a C₂ to C₈ mono-olefin.
- 4) A composition according to either of claims 2 or 3 which further comprises a sorbitol-based clarifier present in any amount between 500ppm and 5000 ppm by weight based on the total weight of said polymer.
- 5) A composition according to either of claims 2, 3, or 4 which further comprises an inorganic clarifier present in any amount between 500 ppm and 5000 ppm by weight based on the total weight of said polymer.
- 6) A composition according to either of claims 2, 3, 4, or 5 which further comprises an inorganic nucleator present in any amount between 250 ppm and 2500 ppm by weight based on the total weight of said polymer.

7) A composition according to any foregoing claim wherein an amine oxide as specified in claim 1 is present, and wherein the ratio of amine oxide to hindered amine light stabilizer is any ratio in the range of between about 1 : 0.2 to 1 : 5.

8) A composition according to any foregoing claim wherein a hydroxyl amine as specified in claim 1 is present, and wherein the ratio of hydroxyl amine to hindered amine light stabilizer is any ratio in the range of between about 1 : 0.5 to 1 : 5 .

Document made available under the Patent Cooperation Treaty (PCT)

International application number: PCT/US04/040366

International filing date: 02 December 2004 (02.12.2004)

Document type: Certified copy of priority document

Document details: Country/Office: US
Number: 60/527,795
Filing date: 08 December 2003 (08.12.2003)

Date of receipt at the International Bureau: 13 January 2005 (13.01.2005)

Remark: Priority document submitted or transmitted to the International Bureau in compliance with Rule 17.1(a) or (b)



World Intellectual Property Organization (WIPO) - Geneva, Switzerland
Organisation Mondiale de la Propriété Intellectuelle (OMPI) - Genève, Suisse